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Investigations on the Influence of Subsequent Electron Beam (EB) Remelting on the Microstructure of an Aluminium Nitride Layer Formed on an Aluminium Substrate (Part II)

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Abstract: Nitriding of Al alloys leads to the formation of a thin, hard nitride layer (AlN) on the surface. A subsequent EBR can both eliminate the nitriding-related cavities under the nitride layer and increase the hardness of the substrate without melting or destroying the nitride layer. This paper deals with investigations regarding the influence of the energy/heat input on the microstructure within both the AlN layer and the remelted Al substrate. Of particular interest was the interface between the AlN and the Al substrate, which changed to a transition zone with a depth of approximately 80 μ m. A range of high-resolution imaging and analytical tools for both scanning and transmission electron microscopy were used for these investigations. Based on the findings from the microstructural investigations, a schematic model was developed of the processes occurring within the nitride layer and at the interface as a result of remelting.

Keywords: plasma nitriding; Al alloy; electron beam; remelting; interface; AlN

1. Introduction

For Fe-based materials, nitriding is a well-known thermochemical surface technology to improve both tribological and corrosive stress behaviour, as well as fatigue strength. This is due to the formation of a thin but very hard compound layer of iron nitrides. In the case of alloyed steels whose alloying elements form special nitrides, this layer is supported by a thicker, hardened diffusion layer.

A hardness or stress profile within the surface similar to that of steel would also be desirable for relatively soft Al-based materials. However, the Al-specific nitriding mechanism—i.e., that of exclusively outward nitride layer growth—leads to strong limitations with respect to the load-bearing capacity of the thin, hard AlN layer formed [1,2]. This is mainly determined by two factors: (i) the lack of a supporting diffusion layer (nitrogen does not exhibit significant solubility in aluminium), and (ii) the formation of local cavities below the AlN layer [1,3,4].

It has already been shown that hardness can be increased by subsequent heat treatment consisting of solution annealing and ageing [5–7], as well as by the duplex treatment of EB surface alloying and nitriding [8]. However, for both technology variants, the cavities cannot be eliminated.

Therefore, a new technology aims at the subsequent electron beam remelting (EBR) of a previously nitrided Al material. As has already been shown, this treatment variant offers completely new possibilities for improving the material's surface properties [9–11]. The basis for this special process sequence is both the significantly higher decomposition (1700 $^{\circ}$ C) and melting temperatures of the AlN (2490 $^{\circ}$ C) [12] than that of the Al base



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). material (<660 °C). This allows melting of the Al substrate surface underneath the already generated AlN layer (1st process) without melting and thus destroying it. Nevertheless, the energy input must be well dosed in order to avoid negatively influencing the AlN layer in terms of both its microstructure and properties. Therefore, the microstructural components of AlN layers as well as their physical properties are of great importance for subsequent surface remelting.

To date, this treatment sequence has only been implemented in association with a subsequent solid-phase process, such as surface hardening for thin nitrided layers or hard coatings deposited on steels [13].

Electron beam liquid-phase processes are characterized by high cooling rates $(10^2-10^5 \text{ K/s} [14])$ after surface remelting, which usually lead to solidification under disequilibrium conditions. In some cases, metastable or completely new phases are precipitated [15]. Furthermore, the solid solution solubility increases [16,17]. In the case of Fe and Zr in Al, the solubility increases from <1 at.% to as much as 10 at.%. For Mg, Si and Cu, as much as 40 at.% can be achieved. However, the information on maximum solubility varies greatly in the literature [18]. Investigations have shown that the rapidly cooled microstructure after spray compacting changes fundamentally after remelting. A dendritic cast structure is formed with eutectic Si precipitates and a supersaturated solid solution. With increasing cooling rate, the Si content in the eutectic can be increased from an original value of about 12 wt.% to as high as 19 wt.% [19]. In addition, new or morphologically altered intermetallic phases are formed [18,20]. This effect strongly depends on the initial chemical composition. Alloying elements such as Cu, Ni and Fe and the intermetallic phases they form have a major influence on the hardness of the remelted layer [8]. In contrast, Si precipitations play something of a subordinate role in changing the hardness of the layer. As a result of microstructural refinement and phase formation, there is typically a 2- to 3-fold increase in hardness after remelting of spray-compacted Al alloys, depending on the cooling rate and chemical composition [11,18].

On the basis of a detailed understanding of the microstructural features of the initial nitrided state presented in part I (see [21]), the changes during subsequent electron beam remelting are investigated in the present work. This applies to both the effects on the AlN layer, with its different phases, as well as the Al base material as a result of short-term remelting, and especially at the interface.

A deep understanding of the mechanisms and microstructural changes evident after the single and combination processes will form the basis for the future generation of a load-optimized surface composite according to the duplex treatment of PN + EBR.

2. Materials and Methods

2.1. Material

The spray-formed hypereutectic Al alloy DISPAL S232[®] (Supplier: Gränges Powder Metallurgy GmbH, Velbert, Germany formerly PEAK Werkstoff GmbH, ibid) was used as the base material. A detailed description of its microstructure can be found in Part I of this research work [21].

2.2. Surface Treatment Technologies and Parameters

Plasma nitriding (1st process step) was carried out in a commercial DC plasma nitriding facility under soft vacuum conditions (15 Pa). The aim of parameter setting (see [21]) was the generation of different nitride layer thicknesses t_{AIN} in the range of 1.5 to 4.3 µm.

In a 2nd process step, EB remelting (EBR) of a small surface area of the Al substrate underneath the already existing AlN layer was carried out in a universal electron beam facility K26-15/80 (power: 15 kW; max. acceleration voltage: 80 kV) from pro-beam systems GmbH (Stollberg, Germany). Due to the process, the treatment takes place in a vacuum chamber with a soft vacuum of approximately 0.1–1 Pa. Therefore, two different beam guidance techniques—line scan and meander—were used (see schematic illustration in [10,11]). In the case of the line scan, the EB is deflected along a line at high frequency

in the y-direction (5000 Hz) and the specimen is fed perpendicularly in the x-direction ($v_x = 8-12 \text{ mm/s}$). This results in a path-shaped energy transfer, the width of which is determined by the line length ($l_y = 10 \text{ mm}$). The energy distribution is almost constant over the line length.

With the meander technique, the EB is guided in a meandering manner over the sample surface by a combined movement of beam deflection in the y-direction and sample displacement in the x-direction. With a relatively slow beam deflection in the y-direction ($f_y = 5-10$ Hz), a discrete melt pool can be seen on the surface compared to the line scan, which is deflected over a path width of b = 10 mm. In the process, the melt tracks created transverse to the sample feed ($v_x = 1.6-2$ mm/s) merge with each other in such a way that a closed remelting path is created. The beam was additionally superimposed by a circular oscillation figure ($d_{osz} = 0.5-0.75$ mm, $f_{osz} = 500-800$ Hz).

The EBR layer thickness t_{EBR} was varied by means of variations of the beam current I_B mainly.

Thermographic measurements (thermo field camera P1-ML (Optris GmbH, Berlin, Germany) on the surface of the AlN layer during EB remelting revealed maximum hotspot temperatures of approximately 1200 °C. These are far below the decomposition or melting temperature of the AlN (cf. Section 1), but they are twice as high as the melting temperature of the Al substrate. As a result, only the Al surface melted up to an energy-dependent depth z.

2.3. Characterization Methods

The process-specific surface deformations after EBR were characterized by non-contact 3D surface profilometry.

The microstructures were analysed by means of metallographic cross sections using light optical microscopy (LOM) and scanning electron microscopy (SEM). In addition to microstructural investigations, the AlN layer thickness t_{AlN} , the EBR layer thickness t_{EBR} and the height difference Δh resulting from surface deformation during remelting were measured on the transverse sections. The above mentioned geometric parameters were determined for each parameter variant on three light microscope images at 2 to 5 measuring points each and then averaged.

SEM images were used to determine the secondary dendrite arm spacing (SDAS). However, the SDAS values were only an estimate, as they were not determined according to recognized standards. With the help of the SDAS, the local solidification time *t* could be estimated according to Equation (1), using the material specific constant $k = 10.4 \,\mu\text{m/s}^{-1/3}$ (for AlSi10Mg [22]). Meanwhile—and considering the solidification interval (T_{s-l}) for DISPAL S232 (216 K [23,24])—the cooling rate \dot{T} was estimated according to Equation (2).

$$SDAS = k \cdot \sqrt[3]{t}$$
 (1)

$$\dot{T} = \frac{T_{s-l}}{t} \tag{2}$$

Transmission electron microscopy JEM 2200FS (JEOL GmbH, Munich, Germany) was used for spatially resolved (AlN layer, interface/transition zone, remelted base material) detailed microstructural characterisation. A comprehensive description of the sample preparation procedures and the high-resolution analytical tools are given already by the authors in Part I [21].

The nanoindentation was performed inside a scanning electron microscope (SEM) using a Picoindenter (PI 87, Hysitron, Minneapolis, MN, USA) mounted on the SEM stage. Hardness-depth curves were recorded in the nitride layer starting from the surface and extending up to the remelted surface of the Al substrate. The indentations were performed using a Berkovich indenter in displacement-controlled mode with a maximum depth of $z_{max} = 60$ nm (nitrided state, PN) and $z_{max} = 80$ nm (nitrided and remelted state,

PN + EBR). The specimen was tiled by 70° relative to the electron beam during the measurement process.

The layers generated in the course of the development of this remelting technology were characterised on the basis of hardness-depth curves by means of microhardness measurement HV0.3 (F_N = 2.98 N).

3. Results and Discussion

3.1. Initial Plasma Nitrided State (PN)

For the preliminary investigations on the influence of the EB parameters on the formation of the EB remelted layer, AlN layers with thicknesses in the range of 1.5 to 5.4 μ m were used (see Section 3.2).

The different layer thicknesses were mainly achieved by varying the effective nitriding time, i.e., the longer the time, the thicker the layer. Detailed information on the nitriding technology and the results observed can be found in the literature [21,25,26].

The main challenges in layer formation as a result of the plasma nitriding of spraycompacted Al alloys are that of local cavity formation due to outward Al diffusion (Figure 1a) and the selective growth of nitride nuclei in the area of the Al solid solution, which led to gap formation at the interface (Figure 1b).



Figure 1. Initial state after plasma nitriding (formation of a thin AlN layer) of a hypereutectic spraycompacted Al alloy with primary silicon Si_P and different intermetallic compounds IC: (**a**) SEM overview image; (**b**) detail from the interface.

The nitrided initial state, which was suitable for detailed investigations by means of high-resolution imaging and analytics after the duplex treatment (PN + EBR) within the context of this contribution, is characterised comprehensively in Part I of this research work [21].

3.2. Influence of the AlN Layer Thickness on the Subsequent Electron Beam Remelting Process (PN + EBR)

Given a constant AlN layer thickness, previous works by the current authors describe the influence of different beam deflection techniques—such as the flash technique [10], the meander technique and the line scan technique [11]—on the microstructure and hardness of the remelted layers.

This contribution investigates the influence of both different AlN layer thicknesses and energy inputs (beam current I_B at a constant feed rate v_x) for two different deflection techniques (line scan, meander).

As expected, the remelting depth increased almost linearly with increasing energy input or increasing beam current I_B (red curve in Figure 2a) with the line scan method. In principle, this relationship also applies to the combination treatment (PN + EBR) if the energy input takes place via the AlN layer formed on the surface. Depending on the initial AlN layer thickness; however, there were deviations with regard to the EBR layer thickness

achieved compared to the single treatment (EBR), i.e., without AlN. Assuming the same energy input, AlN initial layer thicknesses of $t_{AlN} \leq 3 \mu m$ led to lower—and $t_{AlN} \geq 4.5 \mu m$ to higher—EBR layer thicknesses than those achieved by EBR of the unnitrided Al material (cf. Figure 2a). For an AlN initial layer thickness of approximately 4 μm , the EBR layer thicknesses achieved for single and duplex treatment were of a similar magnitude (Figure 2a).



Figure 2. Influence of the initial AlN layer thickness (values given in the diagrams) and the energy input (beam current I_B) on the EBR layer thickness achieved in comparison to the reference EB-remelted Al base material without a nitride layer (EBR) using different beam deflection techniques: (a) line scan ($v_x = 12 \text{ mm/s}$); (b) meander ($v_y = 1.6 \text{ mm/s}$).

With the meander technique, such a clear dependency on both the energy input and the initial AlN layer thickness could not be detected (Figure 2b). The energy input represented by the beam current I_B could only be varied in a narrow range. A beam current of I_B = 1.6 mA or a layer thickness of approximately 50 µm represented a limit state where the AlN layer thickness had a strong influence on the results achieved. While no defects could be found within the EBR layer for thinner AlN layers (cf. area I. for I_B = 1.6 mA, Figure 2b), remelting in association with thicker AlN layers tended to lead to defects, such as layer spalling (cf. area II. for I_B = 1.6 mA, Figure 2b).

During EBR close to the surface, on the one hand, a volume expansion occurred during the generation of the melt pool, which was a function of the thermal expansion coefficient of the Al alloy DISPAL S232 ($\alpha = 18-19 \cdot 10^{-6} \text{ K}^{-1}$) and the temperature difference ($\Delta T = 1100 \text{ K}$). This could be roughly estimated on the basis of the linear expansion perpendicular to the sample surface and amounted to approximately $\Delta z = 1 \mu \text{m}$ for a relevant remelting depth of 50 μm .

On the other hand—and of far greater significance—the local cavities that resulted from the outward diffusion of aluminium during nitriding were filled by the melt (Figure 3b). The temperature- and concentration-dependent melt movement (Marangoni flow) led to swelling of the surface, with some highly geometric patterns at the surface. In the cross section, a relatively uniform wave profile is recognisable, which is characterised by a height difference Δh (cf. Figure 3a). The evaluation of all determined height differences (as described in Section 2.3) showed that the AlN layers are predominantly damage tolerant up to a height difference of $\Delta h_{max} < 16 \ \mu m$, i.e., they can deform and do not lead to spalling on the surface (see exemplarily A in Figure 4a, $t_{AlN} = 1.6 \ \mu m$). Depending on the thickness of the AlN layer and/or the EBR parameters, some spalling of the AlN layer occurred locally (B in Figure 4a, $t_{AlN} = 4.3 \ \mu m$).







Figure 4. Influence of both the initial AlN layer thickness t_{AlN} and the beam current I_B (line scan; $v_x = 12 \text{ mm/s}$) on the surface deformation: (a) $t_{AlN} = 1.6 \mu \text{m}$ at $I_B = 10 \text{ mA}$ (top); $t_{AlN} = 4.3 \mu \text{m}$ at $I_B = 12 \text{ mA}$ (bottom); (b) element distribution after EBR ($I_B = 12 \text{ mA}$) measured by means of cross-sectional analysis using EDS.

EDS measurements proved that even after the combination treatment (PN + EBR), the AlN layer was enriched with Mg and Al (Figure 4b). Initially, this suggested that, compared to the initial AlN layer (see [21]), no changes in either the elemental distribution or, perhaps, the microstructure occurred after EBR as a result of the thermal exposure.

Similarly to the findings shown elsewhere by the authors [9–11,18,20], the rapid solidification of the melt pool generated in the surface of the Al substrate beneath the AlN layer resulted in a hardness level that was up to three times higher than that of the base material. The reason for this was the formation of a new microstructure under non-equilibrium conditions (see needles in SEM micrograph in Figure 4b). This led to the formation of various intermetallic phases, which will be described later.

3.3. High-Resolution Analysis of the Single and Duplex Treatment

3.3.1. Single Treatment: Electron Beam Remelting (EBR)

As already mentioned, the remelted microstructure was so fine—due to the rapid solidification—that the microstructural components could hardly be resolved under the



light microscope. TEM images showed that a cellular dendritic structure and various intermetallic phases (IC) had formed (Figure 5a).

Figure 5. EB remelted microstructure: (**a**) TEM bf image overview with characteristic microstructural areas (IC—intermetallic compounds; CCR—cell core region; ICR—intercellular region); (**b**) detailed image with EDS measuring points (A–D); (**c**) detailed image with SAED measuring areas: I. cell core (CCR), II. intercellular region (ICR); (**d**) EDS analysis (d*–reciprocal lattice plane spacing) of measuring points marked in (**b**); (**e**) radial intensity distribution of the diffraction patterns from area I. and II. marked in (**c**); (**f**) diffraction pattern from <116>-beam direction.

The cells consisted predominantly of Al solid solution. In contrast to the initial spray compacted state (cf. [21]), additionally very fine globular crystallites were seen within the cells (see I. in Figure 5c) and in the intercellular region (II. in Figure 5c) in the bright-field (bf) contrast of the TEM image. In both regions, beside Al additionally Si could be detected by means of EDS measurement (point A, B in Figure 5d). Furthermore, fine-range diffraction confirmed the presence of Si (Figure 5e). The Si crystallites exhibited a size of between 10–20 nm and—in isolated cases—as much as 50 nm (Figure 5c). It was concluded from the diffraction images that the grains were statistically randomly oriented and, thus, produced closed diffraction rings. From the findings, it can be concluded that the silicon that was exclusively precipitated primarily in the initial state is now eutectically precipitated, especially in the intercellular region.

A cell size of 0.15–0.25 μ m was determined on the basis of the TEM images (cf. Figure 5a). It should be noted that this was only an estimate and not a standard-compliant determination of SDAS. A cooling rate of approximately 10^{-3} K/s was calculated according to Equations (1) and (2).

In the analysis range of a few μ m, it could be concluded from the diffraction patterns that the Al matrix within the cells (CCR)—as well as in the intercellular region (ICR)—were of similar orientation. As with a casting texture, the dendrites were oriented towards the heat flux in the direction of the base material. From the <116> beam direction in the TEM, directional solidification resulted parallel to the heat flow near the <110> direction or plane (identical in the cubic crystal system) (see Figure 5f).

Long, acicular $Al_x Fe_y Si_z$ phases contained no or only minor fractions of Cu (cf. analysis point C in Figure 5b,d). Smaller intermetallic phases embedded in the dendritic matrix

microstructure exhibited equal proportions of Fe and Cu alongside the primary fractions of Al and Si (cf. analysis points D in Figure 5b,d).

3.3.2. Duplex Treatment: Plasma Nitriding + Electron Beam Remelting (PN + EBR)

The energy input of the electron beam changed the interface between the nitride layer and the substrate as a result of the rapid melting and solidification during EBR. The cavities resulting from nitriding were filled with molten aluminium (Figure 3). As a result, the clear assignment of the interface in the merely nitrided state transformed into a transition area, the boundaries of which could not be determined precisely on the basis of the microstructure (cf. Figure 6a). An approximately 50 to 200 nm-wide mixing or transition zone of solidified Al melt and AlN was formed.



Figure 6. Analysis at the transition zone between AlN and the EBR layer: (**a**) TEM image (bf) of the area investigated marked with EDS and SAED measuring points (A–H) at the different distances *z*, given in the partial images; (**b**) EDS profiles (note: without N and O); (**c**) EEL spectra measured.

EDS measurements should first provide information about changes in the chemical composition within the transition zone formed. It should be noted that reliable determination of the light elements nitrogen and oxygen is not possible with this method, so that only normalized proportions can be given.

An EDS line scan at the interface EBR layer to the AlN layer (Figure 6a) proved that the Mg enrichment already found in the nitrided initial state also persisted after remelting (Figure 6b). In contrast to the nitrided state [21], a significant Si content was also detectable within the AlN layer (Figure 6b). The Fe and Cu contents were at a comparable, though low level and only occurred locally. By means of EEL spectroscopy, it could be proven—in principle—that both O and N were present in this area. However, no statement could be made about their concentrations.

SAED analysis of different parts (I. lower; II. middle; III. upper; see Figure 7) of the AlN layer after EB treatment revealed that the upper part of the AlN layer (III.) was not affected (Figure 7d,e). Analogous to the merely nitrided initial state previously investigated [21], the EELS spectrum of the upper part showed two strong peaks for AlN and Al (diffusion paths) (Figure 7e). In contrast, a peak for AlN was visible in the middle part of the layer (II., Figure 7c), but it was strongly broadened in the direction of the Si peak (Figure 7e). The spectrum also showed an elevation in the area of the Al peak, though there was no clear maximum, but rather a similar broadening. It was obvious that melting of the Al present in

the initial state (i.e., former Al diffusion paths) occurred here as a result of the energy input. The EEL spectrum in the lower area of the AlN layer (I., Figure 7b) was similar to that of the middle area, with a somewhat clearer Si peak (Figure 7e). It was thus concluded that the Si could rise into the middle part of the AlN layer via the transition region. As a result of the rapid solidification, a fine Al-Si eutectic formed, which led to broadening of the peak.



Figure 7. TEM images from AlN layer after (**a**) plasma nitriding (initial state) and after subsequent EBR at different distances, (**b**) lower part of the AlN layer (transition zone), (**c**) middle part, and (**d**) upper part with marked areas of SAED analysis (I.–III.) and (**e**) corresponding EEL spectra.

Additional fine-range diffraction was carried out in the transition zone between the AlN and the EBR layers (circle area I. (b) in Figure 7b). In the radial intensity distribution, a very high proportion of MgO was detected in addition to AlN and Si (Figure 8c). The intensity distribution of this oxide showed that the MgO was polycrystalline and was present in a preferred orientation (texture). The inner rings of the diffraction pattern were assigned to Si (Figure 8a).



Figure 8. SAED measurements at the transition zone (PN + EBR): diffraction patterns of: (a) MgO, (b) Mg_3N_2 , and (c) radial intensity distribution of the diffraction pattern.

Furthermore, in the transition zone $\{211\}$ reflections of Mg₃N₂ were detected by means of diffraction patterns (Figure 8b). Despite the relatively low decomposition temperature

(approximately 800 °C), the thermal effect of the EB could not decompose the nitride. This can be explained by the fact that the introduced energy of the EB decreases continuously starting from the surface, so that in the transition zone between substrate and EBR layer was a maximum temperature that is lower than the melting temperature of the substrate (516–738 °C).

Since Mg_3N_2 was present next to the molten Al, a reaction to AlN was conceivable.

3.3.3. Characterization of Layer Hardness

Hardness-depth profiles were recorded by means of nanoindentation using a Picoindenter in the SEM. Due to surface rounding during sample preparation, the interface between the AlN and the Al substrate was taken as a reference point, i.e., $z = 0 \mu m$ (Figure 9a). Therefore, hardness measurements in the AlN layer exhibited negative depth values (-z), with positive depth values (z) in the Al substrate. It should be noted that the nanoindentation was carried out under different displacement-controlled test conditions for the treatment states investigated, namely $z_{max} = 60 \text{ nm}$ (PN) and $z_{max} = 80 \text{ nm}$ (PN + EBR). For this reason, the absolute values of the hardness measurements were not directly comparable. The hardness values after the combination treatment PN + EBR normally tend to be underestimated.



Figure 9. Nanoindentation hardness measurements in displacement-controlled mode with a maximum depth of $z_{max} = 60 \text{ nm}$ (PN) and of $z_{max} = 80 \text{ nm}$ (PN + EBR): (**a**) SEM image with indentation tips (exemplary), (**b**) hardness-depth profile (z = 0 marks the transition zone between AlN and EBR layer).

EBR led to nearly four-times-higher hardness (4.6 ± 0.6 GPa) than the initial hardness of the Al substrate. As a result of the thermal influence on the AlN layer during subsequent remelting, there was obviously a homogenisation of the hardness values in the nitride layer (Figure 9b)—values that were strongly scattered in the initial state (cf. [21]). While the average hardness value of the AlN layer after EBR was 13.1 ± 0.6 GPa, the hardness within the transition zone was lower (10.1 ± 1.5 GPa).

3.4. Model Conception for the Formation of the Interface after EB Remelting

On the basis of the TEM investigations in the previous [21] and present contributions, a detailed understanding of the processes involved in nitriding and subsequent remelting was gained. The essential microstructural changes after remelting in comparison to the initial nitrided state were schematically summarised in a model (Figure 10).



Figure 10. TEM images and schematic illustration of the interface (IF) or transition zone (TZ) between the AlN layer and (**a**) the Al substrate in the initial state after plasma nitriding, and (**b**) the remelted microstructure after subsequent EBR (α -Al—Al solid solution; Si_P—primary silicon; Si_E—eutectic silicon; IC—intermetallic compound).

After nitriding an Al alloy, as known, an AlN layer is formed on the surface. For the outward nitriding process, Al diffusion paths (Figure 10a) were detected by means of unbound Al that probably accumulated in the region of the AlN crystallite boundaries [21]. Enrichment with Mg occurred at the interface to the Al substrate, which led to the additional formation of both MgO and Mg₃N₂ [21].

After EBR with suitable parameters, the AlN layer remained fundamentally unaffected. A decisive difference to the merely nitrided state was that the cavities (Figure 10a) formed there had now been filled with melt and, thus, were eliminated after EBR (Figure 10b). Furthermore, the small (but, in the lower-layer section, in particular, wider) gaps between the AlN rods (see Figure 10a) were also filled with Al melt. As a result, a homogeneous transition zone was generated that exhibited continuous bonding of the AlN layer to the substrate. In the lower (transition zone) as well as in the middle part of the AlN layer, additional Al and Si were detected. It was assumed that this formed within the pre-existing Al diffusion paths during the previous nitriding process (cf. Al in Figure 10a). The fine cellular dendritic solidification of the Al melt resulted from the extremely high heating and cooling rates within these small channels. As a result, all of the spaces between the individual AlN rods were infiltrated with melt (due to capillarity) in a quasi-sponge-like manner (Figure 10b).

4. Conclusions

Combination surface treatments have great potential for improving properties that transcend the known possibilities offered by single surface treatments. This was the motivation for the present research work, which dealt with in-depth microstructural investigation of the duplex treatment plasma nitriding and electron beam remelting of a spray-compacted Al alloy. Within the scope of this contribution, the following conclusions could be reached:

- (1) EBR of the Al substrates beneath the AlN layer led to the formation of a fine dendric Al matrix microstructure with different intermetallic compounds.
- (2) In addition to an increase in hardness in the remelted layer, the special innovation of the technology lies in the fact that it was also possible to eliminate (i.e., to "fill") both

the cavities underneath the AlN layer resulting from outward diffusion of the Al and the small gaps between the AlN rods. As a result, a homogeneous transition zone was generated that exhibited continuous bonding of the AlN layer to the substrate.

- (3) Despite the high temperature during the EBR, the transition zone exhibited both AlN and Mg₃N₂. Furthermore, large amounts of MgO were detected in a polycrystalline structure with a preferential orientation. In the lower (transition zone) as well as in the middle part of the AlN layer, additional Al and Si were detected, the radial intensity distribution peaks of which were strongly broadened and washed out. This was obviously due to the fine cellular dendritic solidification of the Al melt.
- (4) The subsequent remelting led to homogenisation of the hardness values in the nitride layer.
- (5) Due to movement of the melt pool, the surface was structured with a relatively regular pattern and defects, both within and below the AlN layer, were filled with Al melt. It is expected that both should have beneficial effects on the material's wear behaviour. The research on this matter has not yet been completed.

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