


Communication

4-Aminoalkyl Quinolin-2-one Derivatives via Knorr Cyclisation of ω -Amino- β -Keto Anilides

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Abstract: In a high-yielding and solvent-free procedure *N*-ethoxycarbonyl protected ω -amino- β -keto anilides undergo Knorr cyclisation in neat polyphosphoric acid to provide straightforward route to 4-aminoalkyl quinolin-2-one derivatives with variable length of the alkyl chain.

Keywords: Knorr; quinolin-2-one; 2-quinolone; carbostyryl; solvent-free

1. Introduction

The quinoline ring system is present in a vast number of natural [1,2] and synthetic [3,4] organic compounds with valuable properties. Among this large group, the subclass of quinolin-2-ones (also known as carbostyryls) stands out with many bioactive structures [5]. For example, the quinolin-2-one fragment is found in alkaloids such as Viridicatin [6–9], Aflaquinolones [10] and Yaequinolones [11], as well as in synthetic drug candidates with anti-inflammatory [12,13] and antibacterial [14] properties. The construction of the quinolin-2-one ring system is most commonly achieved via the classic Knorr cyclisation of β -keto anilides in acidic media [15,16]. The mechanism of this reaction has been studied in detail [17] and also an alternative approach based on *N*-aryl amides of 3-arylpropynoic acids has been developed [18]. In addition to this classical method, the scope of which is limited in the presence of acid-sensitive functionalities, there have been many recent developments. The modern approaches include Pd-catalysed formation of C-C or C-N bonds in the ring system [19–21], Pd-catalyzed synthesis from quinoline *N*-oxides and azodicarboxylates [22], Co-catalyzed cyclization of α -bromo-*N*-phenylacetamides [23], Intermolecular addition/cyclization of carbamoyl radicals under photoredox [24] or Ag [25] catalysis, hypervalent iodine(III)-mediated decarboxylative cyclization [26] and chemoenzymatic approaches [27,28].

Quinolin-2-ones with aminoalkyl substituent at position 4 are interesting as building blocks for complex natural products [29,30] and also in their own right as bioactive substances [12–14]. To date, all instances of these molecules in the literature are synthesised by either S_N2 amination of the corresponding 4-halogenoalkyl derivatives [12,13,31,32] or hydrogenation of the corresponding 4-cyano derivatives [14]—approaches that work mostly for the preparation of 4-aminomethyl derivatives and are not well suited for derivatives with a longer carbon chain between the amino functionality and the quinolin-2-one core. In this communication, we demonstrate that the Knorr reaction can be successfully carried out with *N*-ethoxycarbonyl protected ω -amino- β -keto anilides, leading directly to the corresponding 4-aminoalkyl quinolin-2-one derivatives with variable length of the alkyl chain.

2. Results

The problematic accessibility of ω -amino- β -keto anilides (**1**) by known methods is probably the main reason why these compounds have not been used as precursors to quinolin-2-ones until now. However, since a method developed recently in our laboratory



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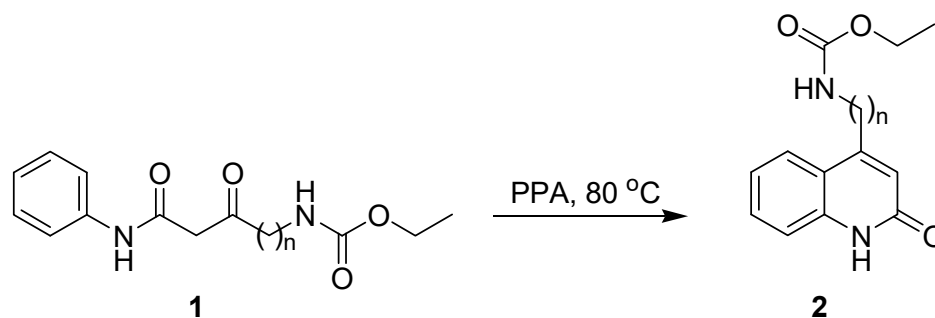
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provided easy access to these substrates [33], we decided to investigate their behaviour under Knorr-type conditions. After a quick screening of various acids and solvents, we arrived at polyphosphoric acid (PPA) as the optimal medium for the targeted cyclocondensation of **1** to 4-aminoalkylquinolin-2-ones **2**. The cyclisation of **1** to **2** (Scheme 1, Table 1) proceeded for 90 min at 80 °C in neat PPA. The products **2** were isolated in 80–90% yield after easy workup, including only the addition of water to the reaction mixture and filtration of the precipitated product or, optionally, extraction in CH₂Cl₂. Although the extractive workup gave slightly cleaner products in case **2b** and **2c**, this synthesis could be carried out as a completely solvent-free procedure, depending on the operator preferences.



Scheme 1. Knorr cyclisation of ω-amino-β-keto anilides to 4-aminoalkylquinolin-2-ones.

Table 1. Yields of 4-aminoalkylquinolin-2-ones **2**, prepared according to Scheme 1.

Product	<i>n</i>	Yield (%)
2a	1	90
2b	2	80
2c	3	85

3. Materials and Methods

The starting *N*-ethoxycarbonyl ω-amino-β-keto anilides (**1**) were prepared from the corresponding ω-amino acids and acetoacetanilide, according to our previously published procedure [33]. Polyphosphoric acid (115% H₃PO₄ basis, CAS No. 8017-16-1) was purchased from (Sigma-Aldrich, Darmstadt, Germany). NMR spectra were run on a Bruker Avance AV600 (600/150 MHz ¹H/¹³C) or Bruker DRX 250 (250/62.5 MHz ¹H/¹³C) spectrometers at BAS-IOCCP—Sofia and chemical shifts (δ, ppm) are downfield from TMS. High resolution mass spectral measurements were performed on a Thermo Scientific Q Exactive hybrid quadrupole-orbitrap mass spectrometer. TLC was conducted on aluminium-backed Silica gel 60 sheets (Merck) with KMnO₄ staining; Melting points were measured on Boetius hot stage apparatus and are not corrected.

Synthetic Procedure

4-aminoalkyl quinolin-2-ones (2a–c), general procedure: To the corresponding β-keto anilide **1a–c** (200 mg) in a glass vial was added PPA (5–6 g, 2.5–3 mL). The mixture was heated to 80 °C and was stirred intensely until full homogenization (ca. 15–20 min). The homogenous mixture was left for a further 90 min. at 80 °C, then the vial was cooled to r.t. with tap water and the contents were rinsed and poured into a glass with 50–70 mL of water. The isolation of the products **2a–c** was conducted by filtration of the resulting suspension (**2a**) or by extraction with 2 × 30 mL CH₂Cl₂ (**2b**, **2c**). The yields of **2b** and **2c** were practically unaffected by the type of workup procedure (filtration or extraction). For product **2a**, filtration is recommended because of its poor solubility in CH₂Cl₂.

(2-Oxo-1,2-dihydro-quinolin-4-ylmethyl)-carbamic acid ethyl ester (**2a**): m.p. 173–174 °C; ¹H NMR (DMSO-*d*₆, δ ppm, *J* Hz): 1.19 (t, *J* = 7, 3H), 4.04 (q, *J* = 7, 2H), 4.42 (d, *J* = 5.9, 2H), 6.32 (s, 1H), 7.18–7.77 (m, 4H, ArH), 7.76 (br t, 1H, NH), 11.71 (br s, 1H, NH); ¹³C NMR

(DMSO- d_6 , δ ppm): 15.1, 41.3, 60.6, 116.1, 118.1, 118.7, 122.2, 124.3, 130.9, 139.3, 148.9, 156.9, 162.1; HRMS (ES⁺): m/z [M + Na]⁺ calcd for C₁₃H₁₄N₂NaO₃⁺: 269.0897, found: 269.0896;

[2-(2-Oxo-1,2-dihydro-quinolin-4-yl)-ethyl]-carbamic acid ethyl ester (**2b**): m.p. 185–186 °C; ¹H NMR (250 MHz, DMSO- d_6 , δ ppm, J Hz): 1.15 (t, 3H, $J = 7$), 2.95 (t, 2H, $J = 7$), 3.29 (m, 2H), 3.98 (q, 2H, $J = 7$), 6.36 (s, 1H), 7.17–7.84 (m, 5H) ArH +NH, 11.64 (s, 1H) NH; ¹³C NMR (DMSO- d_6 , δ ppm): 161.51, 156.31, 148.74, 138.96, 130.16, 124.32, 121.68, 120.99, 118.80, 115.68, 59.60, 39.74, 31.82, 14.62; HRMS (ES⁺): m/z [M + Na]⁺ calcd for C₁₄H₁₆N₂NaO₃⁺: 283.1053, found: 283.1055;

[3-(2-Oxo-1,2-dihydro-quinolin-4-yl)-propyl]-carbamic acid ethyl ester (**2c**): m.p. 116–118 °C; ¹H NMR (250 MHz, CDCl₃, δ ppm, J Hz): 1.27 (t, 3H, $J = 7$), 1.97 (m, 2H), 2.94 (t, 2H, $J = 8$), 3.34 (m, 2H), 4.15 (q, 2H, $J = 7$), 4.98 (br s, 1H) NH, 6.66 (s, 1H), 7.23–7.74 (m, 4H) ArH, 12.67 (br s, 1H) NH; ¹³C NMR (DMSO- d_6 , δ ppm): 164.12, 156.85, 152.89, 138.42, 130.69, 124.02, 122.87, 119.78, 119.04, 117.11, 60.90, 40.63, 29.40, 29.20, 14.66; HRMS (ES⁺): m/z [M + Na]⁺ calcd for C₁₅H₁₈N₂NaO₃⁺: 297.1210, found: 297.1206.

Supplementary Materials: The following are available online, S1.PDF—processed ¹H and ¹³C NMR spectra. S2.zip—Raw NMR data, and mol files structure.

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