

# New Derivatives of Lupeol and Their Biological Activity

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**Abstract:** The natural product lupeol (**1**) was isolated from *Bombax ceiba* leaves, which were used as starting material in the semisynthetic approach. Three new derivatives (**2a**, **2b**, and **3**) were synthesized using oxidation and aldolization. Their chemical structures were elucidated by spectroscopic analyses (HRESIMS and NMR). Compounds **3** showed significant  $\alpha$ -glucosidase inhibition with an  $IC_{50}$  value of 202  $\mu$ M, whereas **2a** and **2b** were inactive.



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**Keywords:** lupeol derivative; benzylidene derivative;  $\alpha$ -glucosidase inhibition; Oxone<sup>®</sup>

## 1. Introduction

Diabetes mellitus (DM) causes high blood glucose after the consumption of a carbohydrate-enriched diet, leading to hyperglycemia. Uncontrolled diabetes is manifested by a very high rise in triglycerides and fatty acid levels [1]. Diverse antidiabetic drugs derived from synthetic compounds are of interest to chemists. However, these synthetic drugs come with several serious complications [1]. Due to the limitations associated with the use of existing synthetic antidiabetic drugs, the search for newer antidiabetic agents from natural sources continues. Lupeol is a pharmacologically active pentacyclic triterpenoid found in several medicinal plants worldwide [2]. It has several potential medicinal properties and is found in a variety of botanical sources [3]. Notably, lupeol has been reported to selectively target diseased and unhealthy human cells, while sparing normal and healthy cells [4]. Dozens of novel lupeol derivatives were synthesized and screened for their in vivo antihyperglycemic activity [5,6]. Most derivatives lowered the blood glucose levels, in a sucrose-challenged streptozotocin-induced diabetic rat (STZ-S) model [5]. To continue our ongoing search for highly efficient antidiabetic agents from derivatized lupeol [6,7], we herein describe the synthesis of lupeol derivatives **2**, **2a**, **2b**, and **3** (Figure 1). The structures of all the obtained compounds were characterized by <sup>1</sup>H, <sup>13</sup>C NMR, and HRESIMS. All derivatives were evaluated for  $\alpha$ -glucosidase inhibition.

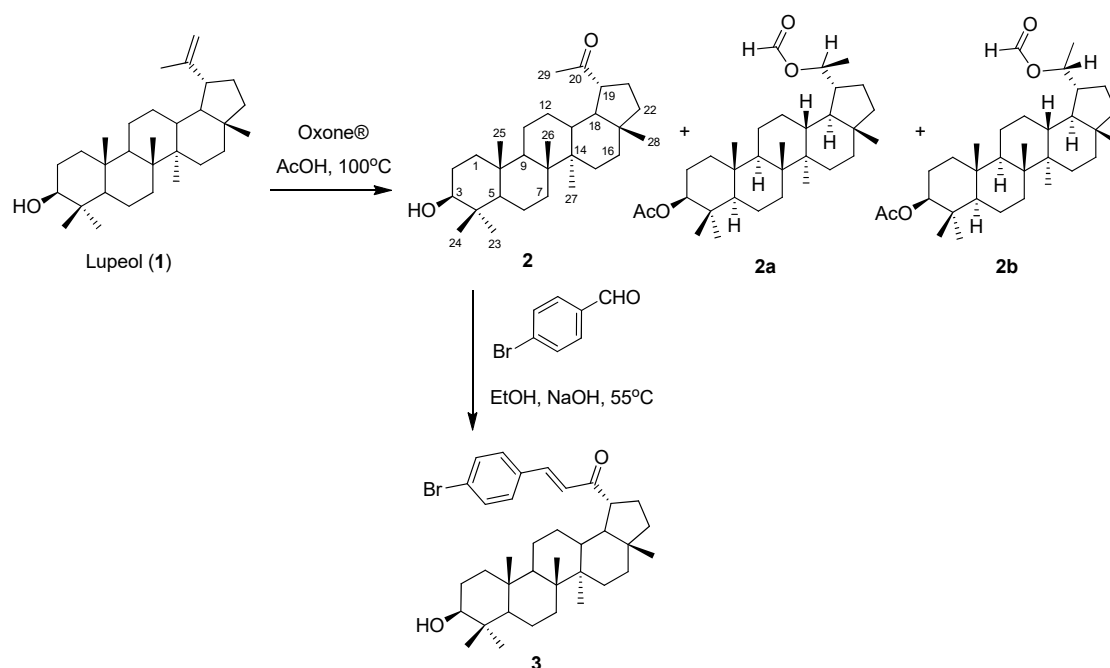


Figure 1. Synthesis of 2, 2a, 2b, and 3 from lupeol (1).

## 2. Results and Discussion

### 2.1. Synthesis

Lupeol was isolated from the Vietnamese plant *Bombax ceiba*, following our previously reported procedure [8]. Lupeol was transformed to products 2, 2a, and 2b using oxidation with Oxone®, a potassium triple-salt ( $\text{KHSO}_5 \cdot 1/2\text{KHSO}_4 \cdot 1/2\text{K}_2\text{SO}_4$ ) [6,9]. The conditions followed our previously reported method [6], with slight modifications. Both 2a and 2b had the same molecular formula as  $\text{C}_{32}\text{H}_{52}\text{O}_4$ . Comparison of NMR data of 2a/2b and 1 indicated that oxidation occurred. The  $^1\text{H}$  NMR spectrum of 2a/2b showed differences with 1: the downfield methine at  $\delta_{\text{H}}$  8.11, two oxymethines at  $\delta_{\text{H}}$  5.26 and 4.48, and a doublet methyl at  $\delta_{\text{H}}$  1.22. These signals indicated that the isopropenyl group of 1 was transformed to a 2-formylethyl group at C-19. Moreover, the downfield signal of H-3 ( $\delta_{\text{H}}$  4.48) indicated that 3-OH was esterified by acetic acid. The  $^{13}\text{C}$  NMR spectrum of 2a/2b showed one carbonyl ester at  $\delta_{\text{C}}$  171.1, one formyl group at  $\delta_{\text{C}}$  163.7 and two oxygenated carbons at  $\delta_{\text{C}}$  81.1 and 72.7, supporting the previous findings. Interestingly, 2a and 2b are C-20 epimers. Corbett and co-workers [10,11] indicated the method to define the absolute configuration of C-20 of lupane-type triterpenes. Particularly, the (20S) and (20R) isomers exhibited differences in the chemical shifts of C-19, C-20, C-29, and C-30, especially C-30. According to Corbett et al., 2a, having C-30 at  $\delta_{\text{C}}$  20.1, would have a 20R configuration. On the other hand, 2b would have the 20S configuration due to the lower chemical shift of C-30 at  $\delta_{\text{C}}$  14.2.

Compound 2 was further aldolized with 4-bromobenzaldehyde to afford compound 3. Compound 3 had the same molecular formula as  $\text{C}_{36}\text{H}_{51}\text{BrO}_2$ , determined by a protonated ion peak at  $m/z$  595.3188 in HRESIMS. Comparison of 1D NMR data of 2 and 3 indicated obvious differences. The first difference is the presence of a 1,4-disubstituted benzenoid characterized by two *ortho*-coupled protons at  $\delta_{\text{H}}$  7.51 and 7.42, and a *trans* double bond at  $\delta_{\text{H}}$  6.75 and 7.46. This was confirmed by the disappearance of a methyl ketone group at  $\delta_{\text{H}}$  2.15 ( $\text{CH}_3$ -29). This finding indicated that the aldolization occurred exclusively at C-29. The second difference was in the  $^{13}\text{C}$  NMR spectrum. This spectrum showed the presence of seven aromatic carbons at  $\delta_{\text{C}}$  141.0 (C-1), 133.9 (C-5'), 132.3 (C-2'), 129.8 (C-3',7'), and 126.9 (C-4', 6'), supporting the reaction at C-29.

## 2.2. $\alpha$ -Glucosidase Inhibition of **2a**, **2b**, and **3**

Compounds **2a**, **2b**, and **3** were evaluated for  $\alpha$ -glucosidase inhibition. Only compound **3** exhibited moderate  $\alpha$ -glucosidase inhibition with an  $IC_{50}$  value of 202  $\mu$ M, compared with an acarbose-positive control ( $IC_{50}$  360  $\mu$ M). Other compounds were inactive.

## 3. Materials and Methods

### 3.1. Materials

Reagents and solvents were obtained from commercial suppliers and were used without further purification. Column chromatography was carried out using Merck Kieselgel 60 silica gel (particle size: 32–63 Å). Analytical TLC was performed using Merck precoated silica gel 60 F-254 sheets.

NMR spectroscopic data were acquired on Bruker Avance III apparatus at 500 MHz for  $^1H$  NMR and 125 MHz for  $^{13}C$  NMR. HRESIMS spectra were recorded on a Bruker MICROTOF-Q 10187.

**Extraction and Isolation.** The air-dried *Bombax ceiba* leaves (4 kg) were ground into powder and exhaustively extracted at room temperature with MeOH (2  $\times$  10 L). The filtered solution was evaporated under reduced pressure to afford a residue (473.4 g). This crude extract was subsequently partitioned using solvents of *n*-hexane and EtOAc to yield *n*-hexane (40 g) and EtOAc (88 g) extracts. The *n*-hexane extract was fractionated by silica gel column chromatography (CC), eluted with *n*-hexane–EtOAc (isocratic, 10:1, *v/v*), to produce five fractions (H1–H5). Fraction H2 (15 g) was rechromatographed by silica gel CC using *n*-hexane–CHCl<sub>3</sub> (isocratic, 12:1, *v/v*) as eluent to afford lupeol (**1**) (1.5 g).

### 3.2. Synthesis Procedure

Synthesis of **2**, **2a**, and **2b**: Lupeol (**1**, 200 mg, 0.469 mmol) was oxidized with Oxone<sup>®</sup> (951 mg, 1.548 mmol) in acetic acid (40 mL) at 100 °C for 3 h. The mixture was stirred and continuously monitored by TLC. The mixture was extracted with EtOAc–water (1:1) to gain the organic layer. This solution was evaporated to afford a residue. Then, the residue was purified by silica gel CC to give compounds **2**, **2a**, and **2b**.

Compound **2**. Isolated yield: 74.6 mg (37%), white solid.  $^1H$  and  $^{13}C$  NMR data were consistent with those reported previously [6].

**(3aR,5aR,5bR,7aR,9S,11aR,11bR,13aR,13bS)-1-((S)-1-(formyloxy)ethyl)-3a,5a,5b,8,8,11a-hexamethylcosahydro-1H-cyclopenta[a]chrysen-9-yl acetate (2a)**. Isolated yield: 9.4 mg (4%), white solid.  $^1H$  NMR (500 MHz, CDCl<sub>3</sub>,  $\delta$ , ppm): 2.05 (3H, *s*, CH<sub>3</sub>-2'), 8.00 (1H, *s*, OCHO-29), 5.33 (1H, *m*, H-20), 4.48 (1H, *dd*, *J* = 10.5, 6.0 Hz, H-3), 2.13 (1H, *m*, H-3), 1.18 (3H, *d*, *J* = 6.5 Hz, CH<sub>3</sub>-30), 1.03 (3H, *s*, CH<sub>3</sub>-26), 0.90 (3H, *s*, CH<sub>3</sub>-27), 0.87 (3H, *s*, CH<sub>3</sub>-25), 0.85 (3H, *s*, CH<sub>3</sub>-23), 0.84 (3H, *s*, CH<sub>3</sub>-24), 0.79 (1H, *d*, *J* = 9.5 Hz, H-5), 0.76 (3H, *s*, CH<sub>3</sub>-28).  $^{13}C$  NMR (125 MHz, CDCl<sub>3</sub>,  $\delta$ , ppm): 171.2 (C-6'), 21.5 (C2'), 161.6 (C-29), 81.1 (C-3), 73.4 (C-20), 55.5 (C-5), 50.2 (C-9), 48.8 (C-18), 43.5 (C-17), 43.0 (C-14), 42.6 (C-19), 41.0 (C-8), 40.5 (C-22), 38.5 (C-4), 38.0 (C-1), 37.3 (C-13), 37.2 (C-10), 35.5 (C-16), 34.4 (C-7), 29.9 (C-21), 28.1 (C-23), 27.3 (C-15), 27.1 (C-2), 23.8 (C-12), 21.0 (C-11), 18.4 (C-6), 18.1 (C-28), 16.7 (C-25), 16.4 (C-26), 16.1 (C-24), 14.4 (C-27), 14.2 (C-30). HRESIMS calcd C<sub>32</sub>H<sub>52</sub>NaO<sub>4</sub> ([M+Na]<sup>+</sup>): 523.3732, found: 523.3763.

**(3aR,5aR,5bR,7aR,9S,11aR,11bR,13aR,13bS)-1-((R)-1-(formyloxy)ethyl)-3a,5a,5b,8,8,11a-hexamethylcosahydro-1H-cyclopenta[a]chrysen-9-yl acetate (2b)**. Yield: 9.4 mg (5%), white solid.  $^1H$  NMR (500 MHz, CDCl<sub>3</sub>,  $\delta$ , ppm): 2.04 (3H, *s*, H-2'), 8.11 (1H, *s*, OCHO-29), 5.26 (1H, *m*, H-20), 4.48 (1H, *dd*, *J* = 11.5, 5.5 Hz, H-3), 2.31 (1H, *m*, H-19), 1.22 (3H, *d*, *J* = 6.5 Hz, CH<sub>3</sub>-30), 1.03 (3H, *s*, CH<sub>3</sub>-26), 0.86 (3H, *s*, CH<sub>3</sub>-25), 0.85 (3H, *s*, CH<sub>3</sub>-27), 0.85 (3H, *s*, CH<sub>3</sub>-23), 0.84 (3H, *s*, CH<sub>3</sub>-24), 0.77 (1H, *d*, *J* = 2.0 Hz, H-5), 0.75 (3H, *s*, CH<sub>3</sub>-28).  $^{13}C$  NMR (125 MHz, CDCl<sub>3</sub>,  $\delta$ , ppm): 171.1 (C-6'), 21.5 (C2'), 163.7 (C-29), 81.1 (C-3), 72.7 (C-20), 55.5 (C-5), 50.0 (C-9), 47.1 (C-18), 44.4 (C-19), 43.2 (C-17), 43.0 (C-14), 41.0 (C-8), 40.1 (C-22), 38.5 (C-4), 38.0 (C-1), 37.5 (C-13), 37.3 (C-10), 35.3 (C-16), 34.4 (C-7), 29.9 (C-21), 28.1 (C-23), 27.4 (C-15), 26.9 (C-2), 23.9 (C-12), 21.0 (C-11), 20.1 (C-30), 18.4 (C-6), 18.1 (C-28), 16.7

(C-25), 16.3 (C-26), 16.1 (C-24), 14.4 (C-27). HRESIMS calcd  $C_{32}H_{52}NaO_4$  ( $[M+Na+H_2O]^+$ ): 541.3870, found: 541.3869.

**Synthesis of 3:** Compound **2** (70 mg, 0.163 mmol) together with NaOH (35 mg, 0.875 mmol) in ethanol (7 mL) was stirred at 55 °C for 15 min. Then, 4-bromobenzaldehyde (64.35 mg, 0.35 mmol) was added to the mixture. The reaction was performed at 55 °C for 2 h. The mixture was extracted with EtOAc–water (1:1, *v/v*) to gain the organic layer. This solution was applied to silica gel CC using the gradient system of *n*-hexane–EtOAc (10:1, *v/v*) to obtain compound **3**. Isolated yield: 68 mg (48%), white solid.

**(E)-3-(4-bromophenyl)-1-((1R,3aR,5aR,5bR,9S,11aR)-9-hydroxy-3a,5a,5b,8,8,11a-hexamethylcosahydro-1H-cyclopenta[a]chrysen-1-yl)prop-2-en-1-one (3):**  $^1H$  NMR (500 MHz,  $CDCl_3$ ,  $\delta$ , ppm): 7.51 (2H, *d*, *J* = 8.5 Hz, H-3',7'), 7.46 (1H, *d*, *J* = 16.0 Hz, H-6'), 7.42 (2H, *d*, *J* = 8.5 Hz, H-4',6'), 6.75 (1H, *d*, *J* = 16.0 Hz, H-29), 3.19 (1H, *dd*, *J* = 11.2, 4.8 Hz, H-3), 2.87 (1H, *td*, *J* = 11.5, 6.0 Hz, H-19), 1.02 (3H, *s*,  $CH_3$ -26), 0.98 (3H, *s*,  $CH_3$ -27), 0.96 (3H, *s*,  $CH_3$ -23), 0.84 (3H, *s*,  $CH_3$ -24), 0.80 (3H, *s*,  $CH_3$ -25), 0.75 (3H, *s*,  $CH_3$ -28).  $^{13}C$  NMR (125 MHz,  $CDCl_3$ ,  $\delta$ , ppm): 204.1 (C-20), 141.0 (C-6'), 133.9 (C-5'), 132.3 (C-3',7'), 129.8 (C-4',6'), 126.9 (C-29), 124.7 (C-2'), 79.0 (C-3), 55.4 (C-5), 50.4 (C-9), 50.1 (C-18), 43.3 (C-17), 42.9 (C-14), 40.9 (C-8), 40.3 (C-22), 39.0 (C-4), 38.8 (C-1), 37.3 (C-10), 35.2 (C-16), 34.3 (C-7), 28.7 (C-15), 28.1 (C-23), 27.9 (C-2), 27.5 (C-12), 21.1 (C-11), 18.4 (C-6), 18.3 (C-28), 16.2 (C-26), 16.0 (C-25), 15.5 (C-24), 14.6 (C-27). HRESIMS calcd  $C_{36}H_{52}BrO_2$  ( $[M-H]^-$ ): 595.3151, found: 595.3188.

### 3.3. $\alpha$ -Glucosidase Inhibitory Assay

The  $\alpha$ -glucosidase (0.2 U/mL) and substrate (5.0 mM *p*-nitrophenyl- $\alpha$ -D-glucopyranoside) were dissolved in 100 mM pH 6.9 sodium phosphate buffer [12]. The inhibitor (50  $\mu$ L) was preincubated with  $\alpha$ -glucosidase; then, the substrate (40  $\mu$ L) was added to the reaction mixture. The enzymatic reaction was carried out at 37 °C for 20 min and stopped by the addition of 0.2 M  $Na_2CO_3$  (130  $\mu$ L). Enzymatic activity was quantified by measuring absorbance at 405 nm. All samples were analyzed in triplicate at five different concentrations around the  $IC_{50}$  values, and the mean values were retained. The inhibition percentage (%) was calculated as follows: Inhibition (%) =  $[1 - (A_{\text{sample}}/A_{\text{control}})] \times 100$ .

## 4. Conclusions

Three new derivatives, **2a**, **2b**, and **3**, from the natural product lupeol have been synthesized via oxidation and aldolization routes and evaluated for their  $\alpha$ -glucosidase inhibition. Synthetic compound **3** showed much stronger  $\alpha$ -glucosidase inhibitory activity ( $IC_{50}$  202  $\mu$ M) than acarbose ( $IC_{50}$  360  $\mu$ M). Synthetic products **2a** and **2b**, which lacked the 3-OH group, exhibited lower activity than **3** toward  $\alpha$ -glucosidase. This result confirmed that this substituted group might be involved in  $\alpha$ -glucosidase inhibition.

**Supplementary Materials:** The following are available online. Copies of HRESIMS and NMR spectra for compound **2a**, **2b**, and **3**.

**Author Contributions:** Conceptualization, T.-H.D. and J.S.; methodology, T.-H.D. and J.S.; formal analysis, H.-T.-T.L., T.-H.D. and J.S.; investigation, T.-H.-T.N., H.-T.-T.L., Q.-C.C., T.-H.D. and Q.-T.-P.T.; data curation, N.-K.-T.P. and N.-H.N.; writing—original draft preparation, T.-H.D. and J.S.; writing—review and editing, T.-H.D. and J.S.; supervision, T.-H.D.; project administration, T.-H.D. All authors have read and agreed to the published version of the manuscript.

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