

Extended Abstract

The Ethaline Effect on Feruloyl Esterase Activity [†]

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In recent years, extractions using natural deep eutectic solvents (NADESs) have attracted increasing attention, as these solvents represent an ecological, non-toxic and biodegradable alternative to conventional solvents, being easy to produce in laboratories [1]. Some studies showed that NADESs considerably increased the extractability of phenolic compounds compared to conventional methodology [1,2]. Another green extraction method is the enzymatic one, which can greatly increase the production of phenolic compounds together with other extraction methods [3]. The purpose of this study was to investigate and optimize the activity of feruloyl esterase (FAE) and the solubility of ferulic acid (FA) in ethaline. The FA solubility was tested at different ethaline concentrations (5%–90%) by the spectrophotometrical method ($\lambda = 320$ nm). The enzyme activity of FAE was optimized by varying the pH values, and the substrate (4-nitrophenyl trans-ferulates) and enzyme (feruloyl esterase) concentrations. After optimizing these parameters, the effect of different concentrations of ethaline (10%–45%) on the enzymatic activity was determined. The optimal pH value for the reaction was 7. The enzyme was still active in ethaline, but the initial velocity was decreased at higher ethaline concentrations, probably due to higher viscosity which decreased the diffusion. The solubility of FA decreased with the decrease in ethaline concentration, but it was soluble enough for FA extractions from biomass, even at ethaline concentrations below 50%. In conclusion, FAE can be combined with ethaline at certain concentrations to improve FA extractions from biomass.

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