

# Supplementary Materials

## Stabilization of chromium (VI) in the presence of iron (II): method development and validation

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### Process water quality

**Table S1.** Analytical data of the process water of the waterworks of the German Environment Agency (UBA) in Berlin-Marienfelde (Germany).

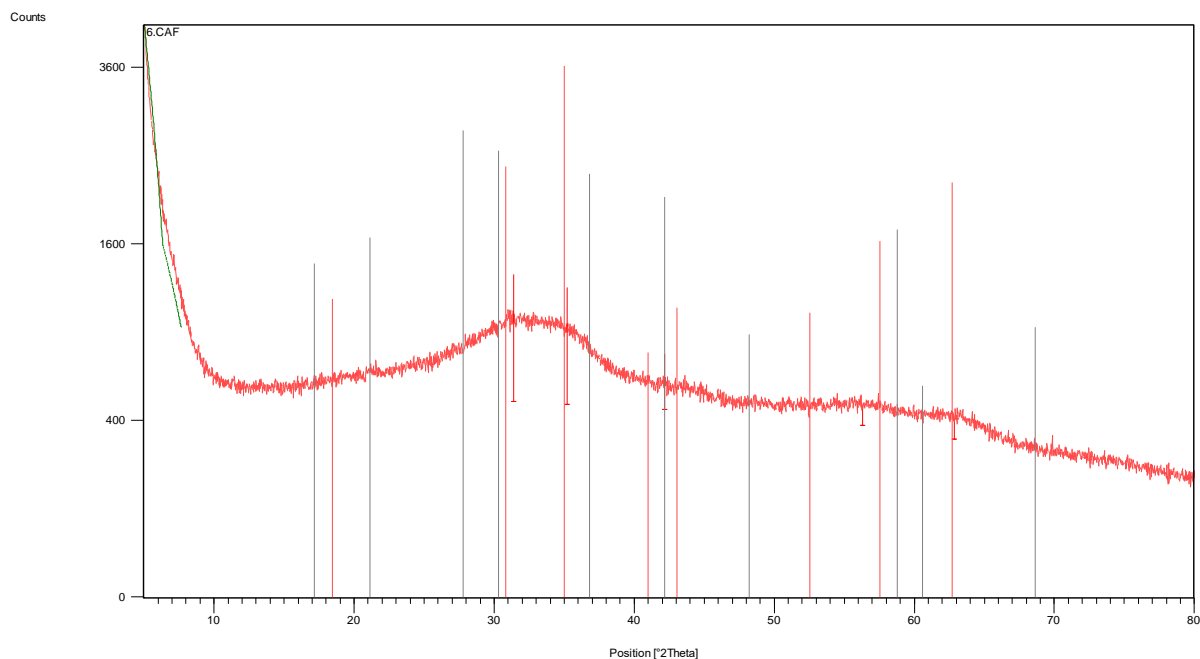
Parameter	Value	Unit
pH at 25 °C	7.56	–
Conductivity at 25 °C	868	$\mu\text{S cm}^{-1}$
Total hardness	20.3	$^{\circ}\text{dH}$
Carbonate hardness	11.2	$^{\circ}\text{dH}$
Carbon dioxide	9	$\text{mg L}^{-1}$
Acid capacity to pH 4.3	4	$\text{mol m}^{-3}$
Base capacity to pH 8.2	0.2	$\text{mol m}^{-3}$
Calcium	126	$\text{mg L}^{-1}$
Magnesium	11.8	$\text{mg L}^{-1}$
Sodium	47.8	$\text{mg L}^{-1}$
Potassium	< 2.0	$\text{mg L}^{-1}$
Silicon	12	$\text{mg L}^{-1}$
Iron	< 0.02	$\text{mg L}^{-1}$
Manganese	< 0.02	$\text{mg L}^{-1}$
Copper	< 0.02	$\text{mg L}^{-1}$
Zinc	0.02	$\text{mg L}^{-1}$
Aluminum	< 0.01	$\text{mg L}^{-1}$
Ammonium	< 0.1	$\text{mg L}^{-1}$
Chloride	75.2	$\text{mg L}^{-1}$
Sulfate	156	$\text{mg L}^{-1}$
Nitrate	< 3.0	$\text{mg L}^{-1}$
Phosphate	< 0.3	$\text{mg L}^{-1}$
Bromide	0.19	$\text{mg L}^{-1}$

## Spearman' rank correlation coefficient

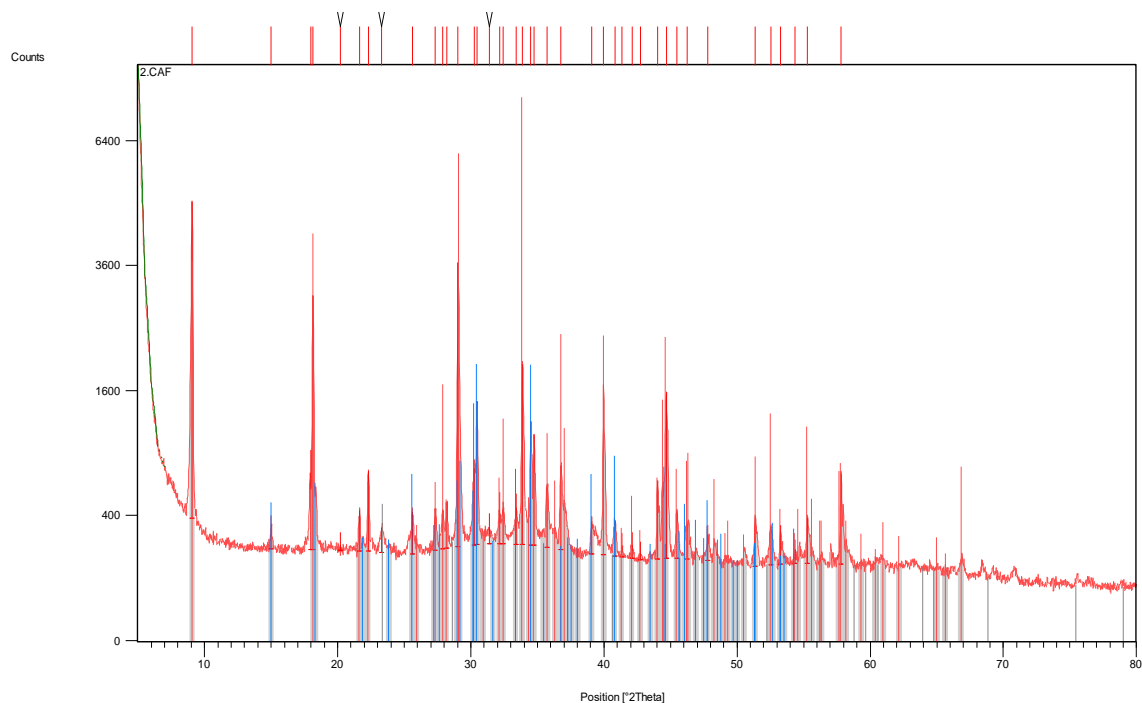
**Table S2.** Correlation between buffer concentration and Cr (VI) recovery.

<b>Buffer System</b>	<b>Fe (II) (mg L<sup>-1</sup>)</b>	<b>pH</b>	<b>Spearman's Rank Correlation Coefficient</b>
HPO <sub>4</sub> <sup>2-</sup>	1	10	1,00
		11	0,95
		12	0,80
	3	10	0,80
		11	0,95
		12	0,80
	6	10	0,50
		11	0,95
		12	0,80
HCO <sub>3</sub> <sup>-</sup>	1	10	0,80
		11	-0,25
		12	-0,40
	3	10	-0,25
		11	-0,85
		12	-0,40
	6	10	1,00
		11	0,80
		12	0,95
NH <sub>4</sub> <sup>+</sup>	1	10	0,80
		11	0,65
		12	0,80
	3	10	0,90
		11	0,10
		12	0,80
	6	10	0,50
		11	0,90
		12	0,90

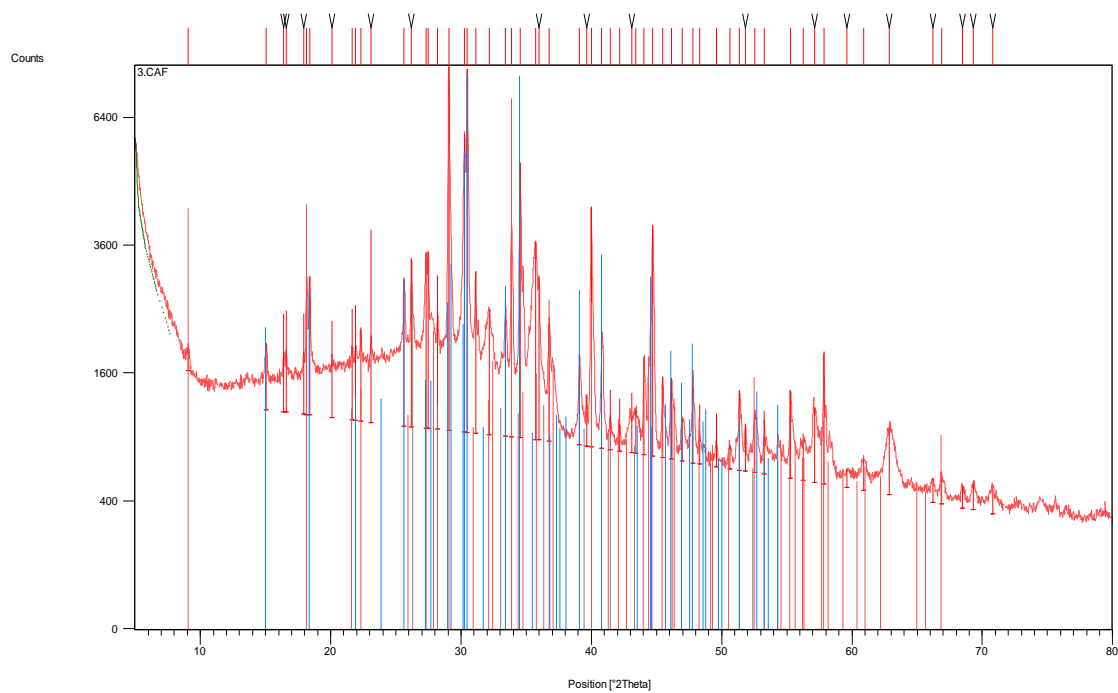
## XRD spectra of the precipitates



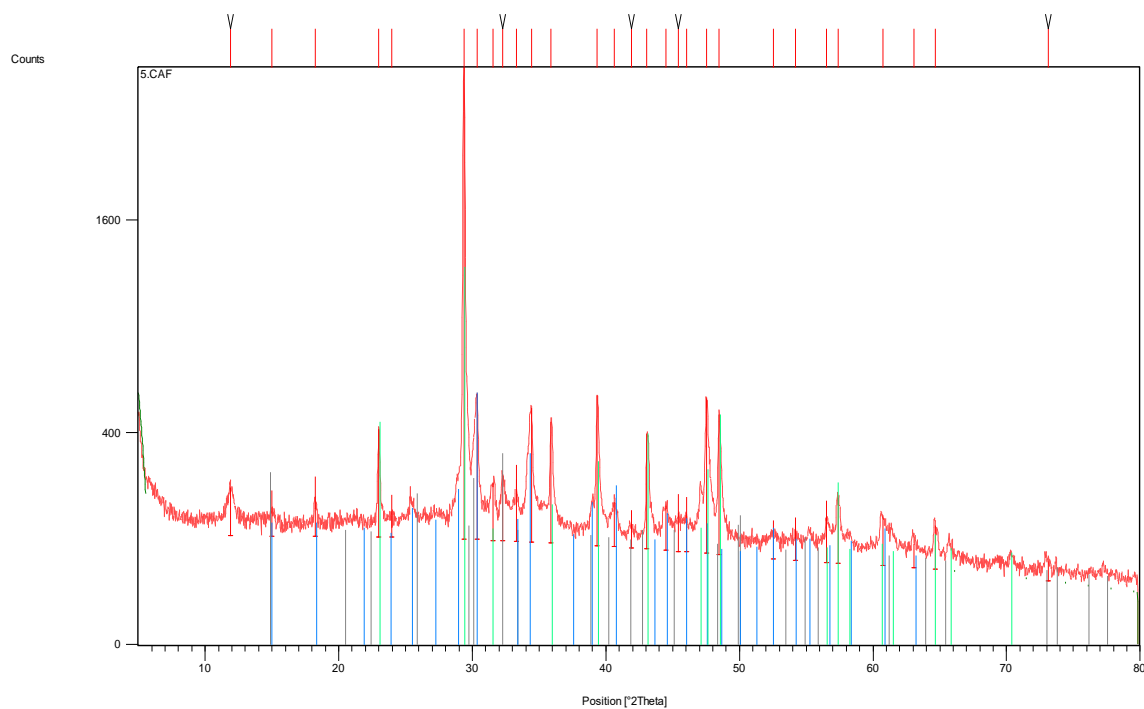
**Figure S1.** XRD spectrum of the precipitates of  $50 \mu\text{g L}^{-1}$  Cr (VI) with  $100 \text{ mg L}^{-1}$  Fe (II) and 30 mM hydrogen phosphate ( $\text{HPO}_4^{2-}$ ) buffer in process water.



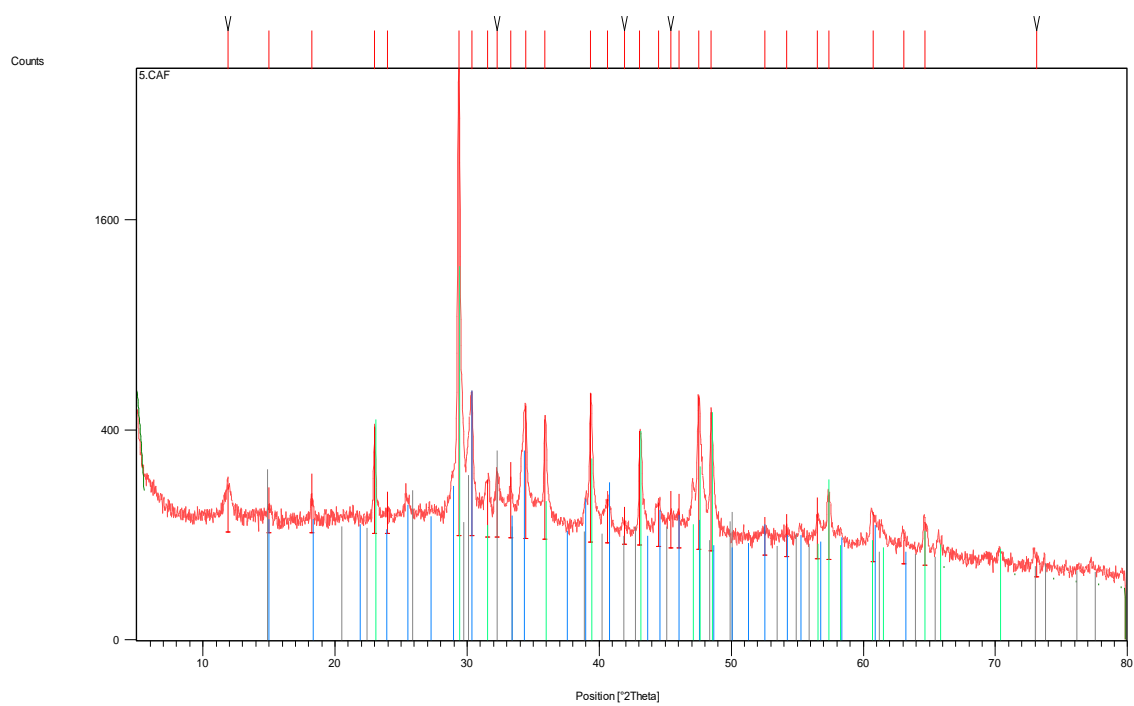
**Figure S2.** XRD spectrum of the precipitates of  $50 \mu\text{g L}^{-1}$  Cr (VI) with  $100 \text{ mg L}^{-1}$  Fe (II) and 30 mM hydrogen carbonate ( $\text{HCO}_3^-$ ) buffer in pure water.



**Figure S3.** XRD spectrum of the precipitates of  $50 \mu\text{g L}^{-1}$  Cr (VI) with  $50 \text{ mg L}^{-1}$  Fe (II) and  $30 \text{ mM}$  hydrogen carbonate ( $\text{HCO}_3^-$ ) buffer in pure water.



**Figure S4.** XRD spectrum of the precipitates of  $50 \mu\text{g L}^{-1}$  Cr (VI) with  $100 \text{ mg L}^{-1}$  Fe (II) and  $30 \text{ mM}$  hydrogen carbonate ( $\text{HCO}_3^-$ ) buffer in process water.



**Figure S5.** XRD spectrum of the precipitates of  $50 \mu\text{g L}^{-1}$  Cr (VI) with  $6 \text{ mg L}^{-1}$  Fe (II) and  $30 \text{ mM}$  hydrogen carbonate ( $\text{HCO}_3^-$ ) buffer in process water.